

SYNTHESIS AND ANTI-HERPES ACTIVITY OF 5-TRIFLUOROVINYL-2'-DEOXYURIDINE

Herdewijn P. *, Kerremans L., Snoeck R., Van Aerschot A., Esmans E. + and De Clercq E.

Laboratory of Pharmaceutical Chemistry and Antiviral Chemotherapy, Rega Institute for Medical Research, Katholieke
Universiteit Leuven, Minderbroedersstraat 10, B-3000 Leuven, Belgium.

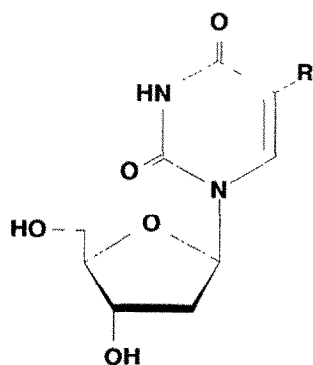
(Received 26 May 1992)

Abstract: 5-Trifluorovinyl-2'-deoxyuridine was synthesized from protected 5-iodo-2'-deoxyuridine and tetrakis(perfluorovinyl)tin. The compound demonstrates significant activity against HSV-1 but low activity against HSV-2.

The introduction of fluorine containing substituents in the 5-position of 2'-deoxyuridine has led to potent antiviral and/or antitumor nucleoside analogues. 5-Fluoro-2'-deoxyuridine (**1**) and 5-trifluoromethyl-2'-deoxyuridine (**2**) are potent inhibitors of deoxythymidylate synthase after anabolic phosphorylation¹. These compounds are used as antitumoral (i.e. 5-fluorouracil) or antiviral (**2**) agents, respectively. 5-Trifluoromethyl-2'-deoxyuridine was synthesized by Heidelberger et al. from 5-trifluoromethyluracil by an enzymatic procedure². Later, methods were described for the direct introduction of a trifluoromethyl group (CF₃I, Cu in HMPA) in the 5-position of pyrimidine nucleosides³. This method proved also useful for the synthesis of the pentafluoroethyl analogue **3**⁴. In contrast to 5-trifluoromethyl-2'-deoxyuridine, however, the pentafluoroethyl analogue **3** is devoid of significant activity against herpes simplex virus type I (HSV-I) and several neoplastic cells⁴.

A second group of interesting fluorinated nucleoside analogues are those derived from (*E*)-5-(2-bromovinyl)-2'-deoxyuridine (BrVdUrd) **4**, which is a nucleoside analogue with high activity and selectivity against both HSV-1 and VZV (varicella zoster virus) replication^{5,6}. It is less active against HSV-2 infections. Several fluorinated analogues of BrVdUrd were synthesized. 5-(2-Fluorovinyl)-2'-deoxyuridine (**5**) was synthesized by Bärwolff from 5-(2-fluoro-2-ethoxycarbonylethyl)-2'-deoxyuridine⁷. The 2-chloro-1,2-difluorovinyl analogues **6** (*E*) and **7** (*Z*) and the 2,2-dichloro-1-fluorovinyl analogue **8** were synthesized by Coe et al. through a sugar-base condensation reaction in low yield⁸. An analogous procedure was used by Bobek et al. for the preparation of 5-(2,2-difluorovinyl)-2'-deoxyuridine (**9**)⁹. In contrast to **6**, **7** and **8**, 5-(2-fluorovinyl)-2'-deoxyuridine **5** and 5-(2,2-difluorovinyl)-2'-deoxyuridine (**9**) demonstrate significant activity against HSV-1 replication (**5**: 1 µg/ml¹⁰; **9**: 0.2 µg/ml⁹). These results prompted us to synthesize 5-(1,2,2-trifluorovinyl)-2'-deoxyuridine (**10**).

Modifications of the original Heck reaction¹¹ have allowed the preparation of a wide variety of unsaturated 5-substituted-2'-deoxyuridine derivatives¹². Recently these reactions are carried out using organoborane or organostannanes as substrates^{13,14}. Based on our experience with symmetric organotin derivatives for carrying out cross-coupling reactions¹⁴, we decided to synthesize the title



1 $R = F$

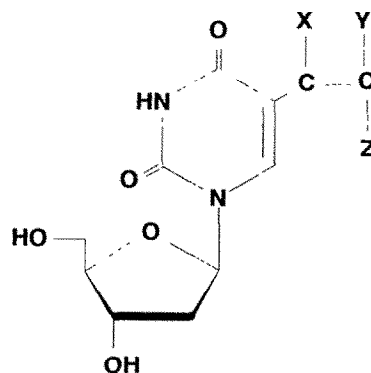
=

2 $R = CF_3$

=

3 $R = CF_2CF_3$

=



	X	Y	Z
4	H	Br	H
=			
5	H	F	H
=			
6	F	Cl	F
=			
7	F	F	Cl
=			
8	F	Cl	Cl
=			
9	H	F	F
=			

compound **10** starting from the appropriately protected 5-iodo-2'-deoxyuridine and tetrakis(perfluorovinyl)tin. Tetrakis(perfluorovinyl)tin was synthesized from tin tetrachloride, magnesium and bromotrifluoroethylene as described by Kaesz *et al* ¹⁵. The reagent was purified by distillation at 23 mbar (bp 55-58°C). Reaction of 3'-O,5'-O,N³-tritoluoyl-5-iodo-2'-deoxyuridine¹⁴ with tetrakis(perfluorovinyl)tin in *N*-methylpyrrolidone in the presence of Pd(OAc)₂, triphenylphosphine and triethylamine afforded the tritoluoyl derivative of **10**. The protecting groups were removed with ammonia in methanol.

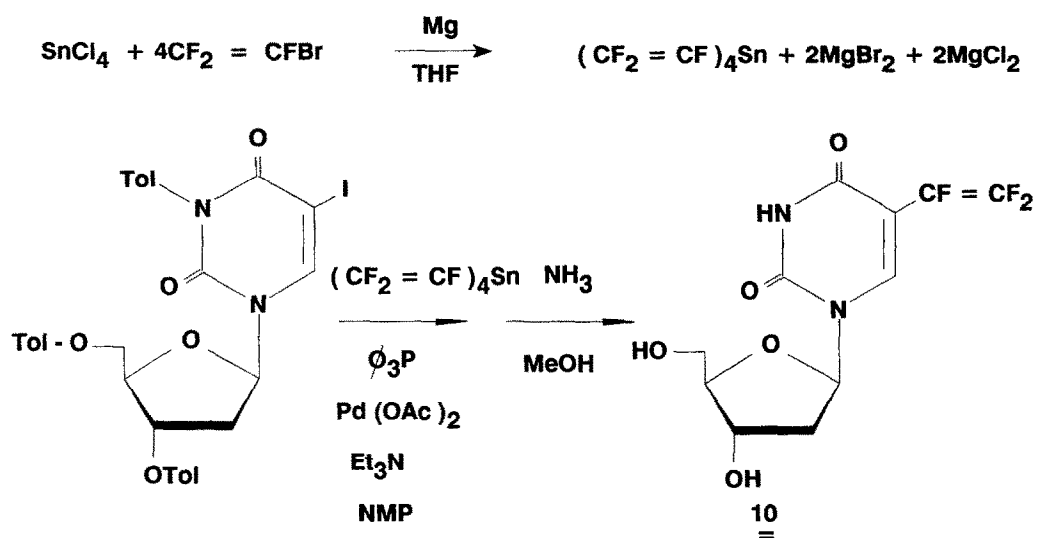


Table 1

Antiviral activity of 5-trifluorovinyl-2'-deoxyuridine in E₆SM cell cultures

Compound	Minimum cytotoxic concentration ^a (μg/mL)	Minimum antiviral concentration ^b (μg/mL)				
		HSV-1 (KOS)	HSV-2 (G)	Vaccinia virus	TK ⁻ HSV-1 (B2006)	HSV-1 (VMW 1837)
10	100	0.05	13	8	13	1.5
4 (BrVdUrd)	200	0.02	100	0.3	100	40

^aRequired to cause a microscopically detectable alteration of normal cell morphology^bRequired to reduce virus-induced cytopathogenicity by 50%

Table 2

Antiviral activity of 5-trifluorovinyl-2'-deoxyuridine in HEL cell cultures

Compound	Minimum cytotoxic concentration ^a ($\mu\text{g/mL}$)	Minimum antiviral concentration ^b ($\mu\text{g/mL}$)			
		VZV		TK ⁻ VZV	
		(OKA)	(YS)	(07/1)	(YS/R)
10	10	0.16	0.55	13.6	11.6
4 (BrVdUrd)	200	0.002	0.004	> 10	> 10

^aRequired to reduce cell growth by 50%^bRequired to reduce virus-induced focus formation by 50%

The activity found for 5-(1,2,2-trifluorovinyl)-2'-deoxyuridine (**10**) against HSV-1 and HSV-2 (Table 1) is comparable to the activity described for the 2,2-difluorovinyl analogue **9**⁹. 5-(1,2,2-Trifluorovinyl)-2'-deoxyuridine is slightly less active than BrVdUrd against HSV-1 but slightly more active than BrVdUrd against HSV-2. The fact that it is less active against TK⁻ than wild-type (TK⁺) HSV-1 points to the importance of the viral thymidine kinase for its intracellular activation by phosphorylation. Also, compound **10** was about 20 to 100 times more active against TK⁺ VZV than TK⁻ VZV (Table 2), again pointing to the importance of the viral thymidine kinase for its activation. Yet, the activity of **10** against TK⁺ VZV is significant lower (100 times) than that of BrVdUrd. While moderate activity against vaccinia virus was observed, 5-(1,2,2-Trifluorovinyl)-2'-deoxyuridine proved to be inactive against vesicular stomatitis virus, poliovirus-1, parainfluenza-3 virus, reovirus-1, Sindbis virus, Coxsackie virus B4 and Semliki forest virus at the highest concentration tested (100 $\mu\text{g/mL}$).

In conclusion, the introduction of a fluorine substituent in the 1 position of the vinyl group does not abolish the anti-herpes activity of 5-(2,2-difluorovinyl)-2'-deoxyuridine. In contrast, substitution of one of the fluorine atoms in the 2 position of **10** by a chlorine atom significantly diminishes the biological activity of **9**⁸.

Experimental

5-(1,2,2-trifluorovinyl)-2'-deoxyuridine (**10**)

A solution of 1.06 g (1.5 mmol) 3'-O,5'-O,N³-tritoluoyl-5-iodo-2'-deoxyuridine¹⁴, 780 mg (3 mmol) triphenylphosphine, 330 mg (1.5 mmol) Pd(OAc)₂ and 2.1 ml (15 mmol) triethylamine in N-methylpyrrolidone (15 ml) was stirred for 10 min at room temperature under nitrogen, after which 1.32 g (3 mmol) of tetrakis(perfluorovinyl)tin was added. The reaction mixture was stirred for 2 hours at room

temperature, poured into H₂O and extracted with Et₂O (twice). The organic layers were combined, dried and dissolved in methanol saturated with ammonia. The mixture was kept for 48 h at room temperature, evaporated, purified by column chromatography (silica gel CH₂Cl₂-MeOH 95:5) and precipitated from acetone. Yield : 100 mg (0,32 mmol, 21 %).

3'-O,5'-O,N³-tritoluoyl-5-(1,2,2-trifluorovinyl)-2'-deoxyuridine : MS CI m/e: 680 (M+NH₄⁺); 328 (BH + NH₄⁺). ¹H NMR (CD₂Cl₂) : 2.35-2.45 (m, H-2', 3xCH₃); 2.84 (dd, H-2''); 4.61 (m, H-4'); 4.67 (dd, H-5'); 4.74 (dd, H-5''); 5.63 (2xt, H-3'); 6.37 (dd, H-1'); 7.30 (m, aromatic-H); 7.79 (d, H-6); 7.92 (m, aromatic-H) ppm.

5-(1,2,2-trifluorovinyl)-2'-deoxyuridine : TLC (CH₂Cl₂-MeOH : 90:10) R_f 0.35. UV λ_{max} = 273 nm. MS CI m/e : 326 (M+NH₄⁺); 309 (M+H⁺); 192 (B+H⁺); 117 (S). ¹³C NMR (DMSO-d₆) : 42.0 (C-2'); 62.3 (C-5'); 71.7 (C-3'); 87.3 (C-4'); 89.2 (C-1') ppm. ¹H NMR (CD₃OD) : 2.05-2.50 (m, H-2', H-2''); 3.78 (m, H-5', H-5''); 3.93 (m, H-4'); 4.40 (m, H-3'); 6.22 (t, J = 6.2 Hz, H-1'); 8.48 (d, J = 2.2 Hz, H-6) ppm. Elem. Anal. (C₁₁H₁₁N₂O₅F₃.C₃H₆O) Calcul. C:45.91%, H:4.68%, N:7.65% found: C:45.61%, H:4.44%, N:7.68%.

The methods used for measuring the inhibitory effects of the compounds on virus-induced cytopathogenicity have been described previously¹⁶.

Acknowledgement

This study was reported by grants from the Belgian Fonds voor Geneeskundig Wetenschappelijk Onderzoek (Projects no. 3.0026.91 and 3.0096.91). A. Van Aerschot is a research associate of the Belgian National Fund for Scientific Research. We thank Dominique Brabants for her dedicated editorial assistance and Anita Van Lierde, Frieda De Meyer and Anita Camps for their efficient technical assistance.

⁺ Laboratory of Organic Chemistry, R.U.C.A., Groenenborgerlaan 171, B-2020 Antwerpen, Belgium.

References

1. Reyes P. Heidelberg C. *Molec. Pharmacol.* **1965**, *1*, 14-30.
2. Heidelberg C. Parsons D.G. Remy D.C. *J. Med. Chem* **1964**, *7*, 1-5.
3. Kobayashi Y., Yamamoto K., Asai T., Nakano M., Kumadahi S. *J. Chem. Soc. Perkin I*, **1980**, 2755-2761.
4. Lin T-S., Gao Y-S. *J. Med. Chem.* **1983**, *26*, 598-601.
5. Walker R.T., Barr P.J., De Clercq E., Descamps J., Jones A.S., Serafinowski P. *Nucleic Acids Res. S.S.* **1978**, n°4, 103-106.
6. De Clercq E., Descamps J., De Somer P., Barr P.J., Jones A.S., Walker R.T. *Proc. Natl. Acad. Sci. USA*, **1981**, *76*, 2947-2951.

7. Bärwolff D., Reefschlager J., Langen P., *Nucleic Acids Res. S.S.* **1981**, n°9, 45-47.
8. Coe P.L., Harnden M.R., Jones A.S., Noble S.A., Walker R.T. *J. Med. Chem.* **1982**, 25, 1329-1334.
9. Bobek M., Kawai I., De Clercq E. *J. Med. Chem.* **1987**, 30, 1494-1497.
10. Reefschläger J., Bärwolff D., Langen P. *Acta Virol.* **1984**, 28, 382-386.
11. Heck R.F. *J. Am. Chem. Soc.* **1968**, 90, 5518-5526.
12. For example: Bergstrom D.E., Ogawa M.K. *J. Am. Chem. Soc.* **1978**, 100, 8106-8112; Dyer R.L., Jones A.S., Walker R.T., Busson R., Vanderhaeghe H. *Nucleic Acids Chemistry* **1991**, part 4, 79-83 Ed. Townsend L.B., Tipson R.S. Wiley Interscience; Whale R.F., Coe P.L., Walker R.T. *Nucleosides and Nucleotides* **1991**, 10, 1615-1624.
13. Crisp G.T., *Synthetic Commun.* **1989**, 19, 2117-2123; Casalnuovo A.L., Calabrese, J.C. *J. Am. Chem. Soc.* **1990**, 112, 4324-4330; Farina V., Hauck S.I., *Synlett* **1991**, 157-159; Wigerinck P., Pannecouque C., Snoeck R., Claes P., De Clercq E., Herdewijn P. *J. Med. Chem.* **1991**, 34, 2383-2389.
14. Herdewijn P., Kerremans L., Wigerinck P., Vandendriessche F., Van Aerschot A. *Tetrahedron Lett.* **1991**, 32, 4397-4400. Mamos P., Van Aerschot, A., Weyns, N., Herdewijn, P. *Tetrahedron Lett.* **1992**, 33, 2413-2416.
15. Kaesz H.D., Stafford S.L., Stone F.G.A. *J. Am. Chem. Soc.*, **82**, 6232-6235.
16. De Clercq E., Descamps J., Verhelst G., Walker R.T., Jones A.S., Torrence P.F., Shugar D. *J. Infect. Dis.*, **1980**, 141, 563-574.